DRAFT

Ashford and Miller (1989. Chemical Sensitivity, a Report to the New Jersey State Department of Health) have summarized studies that were conducted to link multiple chemical sensitivity to exposure to VOCs and other organic compounds.

In a protocol described in a paper by B. Siefert (Regulating Indoor Air, Indoor Air 90 Proceedings of the 5th International Conference on Indoor Air Quality, Vol. 1, Ottawa, 1990) total VOC concentrations indoors greater than $300 \,\mu g/m^3$ are a point of concern to occupants. Total VOC concentration exceeded $300 \,\mu g/m^3$ at 5237 Sun Valley Court in Clarkston, Michigan.

A major difficulty when attempting to identify sources of emissions in buildings is that some porous indoor building materials act as sinks for VOCs from other sources. That is, VOCs from one source may adsorb to the surface of a porous material in the room. When the porous surface is ventilated by clean air, the VOC will be emitted into the room air.

While it is possible that the screening material manufactured by Phifer Wire Products, Inc. and installed in these three homes did contribute to the potential occupants' overall exposure to VOCs, the sampling does not clearly indicate that the screen material is the single or even the major contributor.

At the home located at 6859 Tanglewood Street in Waterford, 12 window screens were placed in a secluded sealed room at an elevated temperature (82°F). This increased temperature would increase the offgassing of compounds into the room air. Results of analysis of samples collected in this room indicated VOC concentrations comparable to the concentrations found in the other two homes and those concentrations noted in the research cited in this section.

This report submitted by:

Ronald C. Poore, IHIT Industrial Hygienist

This report reviewed by:

Stephen D. Paul, CIH Manager, Industrial Hygiene Services Midwestern Operations

April 13, 1993

APPENDIX A

DIRECT-READING MEASUREMENTS

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Direct-Reading Measurements Three Homes in Clarkston and Waterford, Michigan for Phifer Wire Products, Inc. Tuscaloosa, Alabama

Clayton Project No. 45870.00

Measurement Date: January 18, 1993

Location	Military Time	Temperature (F°)	Relative Humidity (%)	Carbon Dioxide (ppm) ^a	Airborne Particulate (µg/m³)b
Outdoors	1015	28	63	300	9
6710 Sun Valley Drive	1030	78.5	26.1	400	20
5237 Sun Valley Court	1352	75.2	22.0	450	10
6859 Tanglewood	1240	73.8	20.5	400	10
Basement storage room 6859 Tanglewood (Temperature maintained for air sampling)	1240	82.5	19.0	400	11
Outdoors	1600	27	61	300	7

parts of carbon dioxide per million parts of air, on a volume-to-volume basis micrograms of particulate per cubic meter of air. on a weight-to-volume basis micrograms of particulate per cubic meter of air, on a weight-to-volume basis

APPENDIX B
EVALUATION CRITERIA

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The following criteria were used to evaluate the indoor air quality in the three homes.

VENTILATION SYSTEM

American Society of Heating, Refrigerating and Air-conditioning Engineers, Inc. (ASHRAE) Standard 62-1989 (Ventilation for Acceptable Indoor Air Quality, Atlanta, 1989) recommends that outdoor air should be supplied to the breathing zone at a flowrate of at least 15 to 20 cubic feet per minute (cfm) per occupant. The occupied space in living areas should be supplied with sufficient outdoor air to assure the air pressure differential is slightly positive relative to the outdoor atmosphere.

CARBON DIOXIDE

Carbon dioxide has been widely used in the industrial hygiene community as a surrogate for various occupant-generated indoor contaminants, such as cigarette smoke and bioeffluents, that are known to cause annoyance, irritation, or discomfort among building occupants (P. Morey, W. Jones, J. Clere, et al., "Studies on Sources of Airborne Microorganisms and on Indoor Air Quality in a Large Office Building." IAQ '86 Managing Indoor Air for Health and Energy Conservation. ASHRAE, Atlanta, 1986, pp. 500-509).

For buildings or zones of buildings where air cleaning to remove smoke and bioeffluents is not provided, performance criteria establishing a limit of 1,000 parts per million (ppm) carbon dioxide (equivalent to about 15 to 20 cfm outdoor air per occupant) are recommended to satisfy occupant comfort needs.

AIRBORNE PARTICULATE

ASHRAE Standard 62-1989 recommends that levels of total particulate in outdoor air entering a building not exceed the United States Environmental Protection Agency (USEPA) National Ambient Air Quality Standards (NAAQS) of 75 micrograms per cubic meter (µg/m³). Air containing higher levels of particulate must be filtered to bring particulate levels within guidelines.

An addendum to Standard 62-1989 (Addendum to Ventilation for Acceptable Indoor Air Quality) recommends that, in accordance with the Clean Air Act of 1987, levels of respirable particulate (particles of less than 10 micron aerodynamic diameter) not exceed 50 µg/m³ in outdoor air entering a building.

Clayton has set maximum indoor air concentration guidelines for total suspended particulate at $75 \,\mu\text{g/m}^3$ and for respirable particulate at $50 \,\mu\text{g/m}^3$. These evaluation criteria reflect our view that performance guidelines for indoor air should at a minimum comply with provisions of the Clean Air Act (USEPA-NAAQS) for outdoor air.

DRAFT

THERMAL ENVIRONMENTAL PARAMETERS

ASHRAE Standard 55-1992R (Thermal Environmental Conditions for Human Occupancy, Atlanta, 1992) defines the environmental conditions that most (90 percent or more) sedentary occupants in indoor environments will find satisfactory. Performance criteria that are established in this standard include the following recommendations for operative temperatures in the occupied spaces: 73 to 78.5°F at 60 percent relative humidity (summertime conditions); 69°F to 76°F at 30% relative humidity (wintertime conditions).

Clayton considers the broad range of relative humidity from 30 to 60 percent as most acceptable for occupant comfort in office environments. Wintertime relative humidities in the 20 to 30 percent range are marginally acceptable but may cause discomfort to sensitive individuals. Humidities below 20 percent are unacceptable because of the effect of dry air on both the mucous membranes and the skin (P. Morey and J. Woods "Indoor Air Quality in Health Care Facilities." Occupational Medicine: State of the Art Reviews 2, 547-563, 1987).

To provide a comfortable living environment during the summer, relative humidity should not exceed 60 percent. Relative humidities greater than 60 percent and certainly those above 70 percent are conducive to the proliferation of microbial agents, such as fungi. Clayton considers the upper and lower dew point limits of 62 and 35°F, respectively, acceptable based on considerations of thermal comfort, respiratory health, and mold growth.

APPENDIX C RESULTS OF SAMPLING AND ANALYSIS

Table 1
Results of Air Sampling and Analysis for Amines at
Three Homes
in
Clarkston and Waterford, Michigan
for
Phifer Wire Products, Inc.
Tuscaloosa, Alabama

Clayton Project No. 45870.00

Sampling Date: January 18, 1993

Sample Number	Description	Samplir Start	g Period Stop	Volume (Liters)	Ethyl amine (ppm)	Diethyl amine (ppm)	Methyl amine (ppm)	Dimethyl amine (ppm)
PH-2	6710 Sun Valley Drive Clarkston, Michigan	1025	1600	67.0	<0.02	<0.03	<0.05	<0.02
PH-6	5237 Sun Valley Court Clarkston, Michigan	1105	1620	62.0	<0.03	<0.03	<0.05	<0.03
PH-8	6859 Tanglewood Street Waterford, Michigan	1230	1645	43.4	<0.04	<0.05	<0.07	<0.04

ppm means parts of contaminant gas or vapor per million parts of air, on a volume-to-volume basis < means "less than"

Table 2
Results of Air Sampling and Analysis for Inorganic Acids at

Three Selected Homes
in
Clarkston and Waterford, Michigan
for
Phifer Wire Products, Inc.
Tuscaloosa, Alabama

Clayton Project No. 45870.00

Sampling Date: January 18, 1993

Sample Number	Description	Samplin Start	g Period Stop	Volume (Liters)	Hydrogen fluoride (ppm)	Hydrogen bromide (ppm)	Hydrogen Chloride (ppm)	Nitric acid (ppm)	Phosphoric acid (ppm)
РН-3	6710 Sun Valley Drive Clarkston, Michigan	1025	1600	67.4	<0.05	<0.06	<0.13	<0.08.	<0.099
PH-4	5237 Sun Valley Court Clarkston, Michigan	1105	1620	65.1	<0.05	<0.06	<0.13	<0.08	<0.0099
PH-9	6859 Tanglewood Street Waterford, Michigan	1230	1645	51.0	<0.05	<0.06	<0.13	<0.08	<0.0099

ppm means parts of contaminant gas or vapor per million parts of air, on a volume-to-volume basis < means "less than"



Sample Matrix/Media: Tenax Tubes Date Sampled: 1/18/93
Analytical Method: Thermal Desorption, GC/MS Date Received: 1/21/93
Lab Number: 179223 Date Analyzed: 1/27/93
Sample Identification: PH-1 Air Volume: 100 Lab

Notorila Compounda	Conce	ntration b	Limit of Detection
Volatile Compounds	<u>Conce</u> (μg)	(μg/m ³)	<u>retection</u> (μg)
Acetone	<0.1	<1	0.1
Benzene	0.72	7.2	0.01
Bromodichloromethane	< 0.01	<0.1	0.01
Bromoform	<0.01	<0.1	0.01
2-Butanone	<0.1	<1	0.1
Carbon Disulfide	< 0.01	<0.1	0.01
Carbon Tetrachloride	< 0.01	<0.1	0.01
Chlorobenzene	< 0.01	<0.1	0.01
2-Chloroethylvinyl Ether	< 0.01	<0.1	0.01
Chloroform	< 0.01	<0.1	0.01
Dibromochloromethane	< 0.01	<0.1	0.01
1,1-Dichloroethane	< 0.01	<0.1	0.01
1,2-Dichloroethane	< 0.01	< 0.1	0.01
1,1-Dichloroethene	< 0.01	< 0.1	0.01
1,2-Dichloroethene (total)	< 0.01	< 0.1	0.01
1,2-Dichloropropane	< 0.01	<0.1	0.01
cis-1,3-Dichloropropene	< 0.01	< 0.1	0.01
trans-1,3-Dichloropropene	< 0.01	< 0.1	0.01
Ethylbenzene	0.33	3.3	0.01
2-Hexanone	< 0.05	<0.5	0.05
Methylene Chloride	<0.1	<1	0.1
4-Methyl-2-Pentanone	<0.05	<0.5	0.05
Styrene	0.69	6.9	0.01
1,1,2,2-Tetrachloroethane	< 0.01	<0.1	0.01
Tetrachloroethene	0.03	0.3	0.01
Toluene	0.76	7.6	0.01
1,1,1-Trichloroethane	< 0.01	<0.1	0.01
1,1,2-Trichloroethane	<0.01	<0.1	0.01
Trichloroethene	<0.01	<0.1	0.01
	<0.01	<0.5	0.05
Vinyl Acetate Xylenes	1.2	12	0.01



Sample Matrix/Media: Tenax Tubes Date Sampled: 1/18/93
Analytical Method: Thermal Desorption, GC/MS Date Received: 1/21/93
Lab Number: 179223 Date Analyzed: 1/27/93
Sample Identification: PH-1 Air Volume: 100 L a

Compound	Conce	ntrationb	
	(μg)	(μg/m ³)	
Tentatively Identified Compounds ^C			
C10 Aliphatic Hydrocarbons	0.51	5.1	
C9 Aromatic Hydrocarbons	2.5	25	
C10 Aromatic Hydrocarbons	0.02	0.2	
Limonene	0.92	9.2	
Butyl Acetate	0.65	6.5	
Hexamethyl Cyclotrisiloxane	0.04	0.4	
Benzofuran	0.27	2.7	

^aAirborne concentrations are based on the air volume provided.

bResults have been blank corrected.

CTentatively Identified Compound results are based on nearest internal standard.

Sample Matrix/Media: Analytical Method: Lab Number:

Tenax Tubes
Thermal Desorption, GC/MS

Date Sampled: 1/18/93 Date Received: 1/21/93 Date Analyzed: 1/27/93 Air Volume: 100 La

Sample Identification:

179224 PH-5

Volatile Compounds	Conce	Concentration ^b		
-	(μg)	(μg/m ³)	Detection (μg)	
Acetone	<0.1	<1	0.1	
Benzene	0.05	0.5	0.01	
Bromodichloromethane	< 0.01	< 0.1	0.01	
Bromoform	< 0.01	<0.1	0.01	
2-Butanone	< 0.1	<1	0.1	
Carbon Disulfide	< 0.01	< 0.1	0.01	
Carbon Tetrachloride	< 0.01	< 0.1	0.01	
Chlorobenzene	< 0.01	< 0.1	0.01	
2-Chloroethylvinyl Ether	< 0.01	< 0.1	0.01	
Chloroform	< 0.01	< 0.1	0.01	
Dibromochloromethane	< 0.01	< 0.1	0.01	
1,1-Dichloroethane	< 0.01	<0.1	0.01	
1.2-Dichloroethane	< 0.01	< 0.1	0.01	
1.1-Dichloroethene	< 0.01	< 0.1	0.01	
1,2-Dichloroethene (total)	< 0.01	< 0.1	0.01	
1,2-Dichloropropane	< 0.01	< 0.1	0.01	
cis-1,3-Dichloropropene	< 0.01	< 0.1	0.01	
rans-1,3-Dichloropropene	< 0.01	< 0.1	0.01	
Ethylbenzene	0.42	4.2	0.01	
2-Hexanone	<0.05	<0.5	0.05	
Methylene Chloride	< 0.1	<1	0.1	
4-Methyl-2-Pentanone	< 0.05	<0.5	0.05	
Styrene	0.38	3.8	0.01	
1,1,2,2-Tetrachloroethane	< 0.01	<0.1	0.01	
Tetrachloroethene	< 0.01	< 0.1	0.01	
Toluene	3.6	36	0.01	
1,1,1-Trichloroethane	30	300 .	0.01	
1,1,2-Trichloroethane	< 0.01	< 0.1	0.01	
Trichloroethene	0.20	2.0	0.01	
Vinyl Acetate	<0.05	<0.5	0.05	
Xylenes .	1.7	17	0.01	

Tenax Tubes Sample Matrix/Media: Date Sampled: 1/18/93 Analytical Method: Thermal Desorption, GC/MS Date Received: 1/21/93 Date Analyzed: Lab Number: 179224 1/27/93 Air Volume: Sample Identification: PH-5 100 La

Compound	Conce	ntration ^b	
	(μg)	(μg/m ³)	
Tentatively Identified Compounds ^C			
C5 Aliphatic Hydrocarbons	0.09	0.9	
C8 Aliphatic Hydrocarbons	0.12	1.2	
C10 Aliphatic Hydrocarbons	0.15	1.5	
C10 Aliphatic Hydrocarbons C11 Aliphatic Hydrocarbons	0.03	0.3	
C9 Aromatic Hydrocarbons	0.12	1.2	
Limonene	0.13	1.3	
Pyrrolidine	0.03	0.3	

aAirborne concentrations are based on the air volume provided.

bResults have been blank corrected.

CTentatively Identified Compound results are based on nearest internal standard.

Sample Matrix/Media: Analytical Method: Lab Number:

Tenax Tubes

1/18/93

Thermal Desorption, GC/MS 179225

Date Sampled: Date Received: Date Analyzed: 1/21/93 1/27/93

Sample Identification:

PH-7

Air Volume: 54 L a

Volatile Compounds	Concentration b		Limit of Detection	
Volatile Componiats	<u>(μg)</u>	(μg/m ³)	<u>εκτασι</u> (μg)	
	<u> </u>	(ug/til-)	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	
Acetone	<0.1	4	0.1	
Benzene	0.09	⊘ 2	0.01	
Bromodichloromethane	< 0.01	<0.2	0.01	
Bromoform	< 0.01	<0.2	0.01	
2-Butanone	<0.1	<2	0.1	
Carbon Disulfide	< 0.01	< 0.2	0.01	
Carbon Tetrachloride	< 0.01	<0.2	0.01	
Chlorobenzene	< 0.01	< 0.2	0.01	
2-Chloroethylvinyl Ether	< 0.01	< 0.2	0.01	
Chloroform	< 0.01	< 0.2	0.01	
Dibromochloromethane	< 0.01	< 0.2	0.01	
1.1-Dichloroethane	< 0.01	< 0.2	0.01	
1,2-Dichloroethane	< 0.01	< 0.2	0.01	
1,1-Dichloroethene	< 0.01	<0.2	0.01	
1,2-Dichloroethene (total)	< 0.01	< 0.2	0.01	
1.2-Dichloropropane	< 0.01	< 0.2	0.01	
cis-1,3-Dichloropropene	< 0.01	< 0.2	0.01	
trans-1,3-Dichloropropene	< 0.01	<0.2	0.01	
Ethylbenzene	0.09	2	0.01	
2-Hexanone	< 0.05	< 0.9	0.05	
Methylene Chloride	<0.1	<2	0.1	
4-Methyl-2-Pentanone	< 0.05	< 0.9	0.05	
Styrene	0.06	1	0.01	
1,1,2,2-Tetrachloroethane	<0.01	<0.2	0.01	
Tetrachloroethene	<0.01	<0.2	0.01	
Toluene	0.42	7.8	0.01	
1,1,1-Trichloroethane	0.01	0.2	0.01	
1,1,2-Trichloroethane	<0.01	<0.2	0.01	
Trichloroethene	<0.01	<0.2	0.01	
Vinyl Acetate	< 0.05	<0.9	0.05	
Xylenes .	0.40	7.4	0.01	

Sample Matrix/Media: Tenax Tubes
Analytical Method: Thermal Desorption, GC/MS
Lab Number: 179225
Sample Identification: PH-7

Date Sampled: 1/18/93
Date Received: 1/21/93
Date Analyzed: 1/27/93
Air Volume: 54 L a

Compound	Conce	ntrationb	
	(μg)	(μg/m ³)	
Tentatively Identified Compounds ^C			
C6 Aliphatic Hydrocarbons	0.05	0.9	
C7 Aliphatic Hydrocarbons	0.33	6.1	
C9 Aliphatic Hydrocarbons	0.13	2.4	
C10 Aliphatic Hydrocarbons	0.17	3.1	
C11 Aliphatic Hydrocarbons	0.07	1.3	
C13 Aliphatic Hydrocarbons	0.54	10.	
C9 Aromatic Hydrocarbons	0.41	7.6	
C10 Aromatic Hydrocarbons	0.15	2.8	
Limonene	0.45	8.3	
Hexamethyl Cyclotrisiloxane	0.07	1	
Octamethyl Cyclotrisiloxane	0.02	0.3	

^aAirborne concentrations are based on the air volume provided.

bResults have been blank corrected.

CTentatively Identified Compound results are based on nearest internal standard.

Sample Matrix/Media: Analytical Method: Lab Number:

Tenax Tubes
Thermal Desorption, GC/MS
179226

Date Sampled: Date Received: Date Analyzed: Air Volume:

1/18/93 1/21/93 1/27/93

Sample Identification:

PH-10 Blank

		_	Limit of
Volarile Compounds	Conce	ntration ^b	Detection
<u>-</u>	(μg)	(µg/m³)	(μg)
Acetone	0.5		0.1
Benzene	< 0.01	••	0.01
Bromodichloromethane	< 0.01		0.01
Bromoform	<0.01	••	0.01
2-Butanone	< 0.1	••	0.1
Carbon Disulfide	< 0.01	••	0.01
Carbon Tetrachloride	< 0.01		0.01
Chlorobenzene	< 0.01	••	0.01
2-Chloroethylvinyl Ether	< 0.01		0.01
Chloroform	< 0.01	••	0.01
Dibromochloromethane	< 0.01		0.01
1,1-Dichloroethane	< 0.01	••	0.01
1,2-Dichloroethane	< 0.01	••	0.01
1,1-Dichloroethene	< 0.01		0.01
1,2-Dichloroethene (total)	< 0.01	••	0.01
1,2-Dichloropropane	< 0.01		0.01
cis-1,3-Dichloropropene	< 0.01	••	0.01
trans-1,3-Dichloropropene	< 0.01		0.01
Ethylbenzene	< 0.01	••	0.01
2-Hexanone	< 0.05	••	0.05
Methylene Chloride	< 0.1	••	0.1
4-Methyl-2-Pentanone	< 0.05	••	0.05
Styrene	< 0.01	••	0.01
1,1,2,2-Tetrachloroethane	< 0.01	••	0.01
Tetrachloroethene	<0.01	••	0.01
Toluene	0.03	••	0.01
1,1,1-Trichloroethane	< 0.01	••	0.01
1,1,2-Trichloroethane	< 0.01	••	0.01
Trichloroethene	< 0.01	••	0.01
Vinyl Acetate	< 0.05	••	0.05
Xylenes	0.01		0.01
•			

APPENDIX D

RESULTS
OF
LABORATORY GAS CHROMATOGRAPHY/MASS SPECTROSCOPY
(GC/MS) STUDY

Weights and Dimensions for Phifer Wire Products

Clayton Project No. 46431-17

Sample Matrix/Media: Screens Date Received: 02/08/93 Analytical Method: TO1 (Modified)

Sample no.	Temperature (C)	Area (cm2)	Weight (ug)
126823	30	17.25	0.192
	50	15.81	0.162
126823	100	5.04	0.055
126823	100	3.04	0.033
126824	30	15.75	0.193
126824	50	11.07	0.143
126824	100	8.84	0.103
126825	30	17.92	0.241
126825	50	17.10	0.210
126825	100	10.34	0.131
120025	100	20.04	3.232
126826	30	20.00	0.250
126826	50	26.00	0.320
126826	100	8.33	0.106

Clayton Project No. 46431-17

Sample Matrix/Media: Screens Date Prepared: 03/04/93
Lab Number: >C9377 Date Analyzed: 03/04/93

Sample Identification: LAB BLANK

Analytical Method: EPA T01 (Modified)

Volatile Compounds	Concentration	Limit o Detecti
	(ug)	(ug)
Acetone	<0.1	0.1
Benzene	<0.01	0.01
Bromodichloromethane	<0.01	0.01
Bromoform	<0.01	0.01
2-Butanone	<0.1	0.1
Carbon disulfide	<0.01	0.01
Carbon tetrachloride	<0.01	0.01
Chlorobenzene	<0.01	0.01
Chloroform	<0.01	0.01
Dibromochloromethane	<0.01	0.01
1,1-Dichloroethane	<0.01	0.01
1,2-Dichloroethane	<0.01	0.01
1,1-Dichloroethene	<0.01	0.01
1,2-Dichloroethene (total)	<0.01	0.01
1,2-Dichloropropane	<0.01	0.01
cis-1,3-Dichloropropene	<0.01	0.01
trans-1,3-Dichloropropene	<0.01	0.01
Ethylbenzene	<0.01	0.01
2-Hexanone	<0.05	0.05
Methylene chloride	<0.05	0.05
4-Methyl-2-pentanone	<0.05	0.05
Styrene	<0.01	0.01
1,1,2,2-Tetrachloroethane	<0.01	0.01
Tetrachloroethene	<0.01	0.01
Toluene	<0.01	0.01
1,1,1-Trichloroethane	<0.01	0.01
1,1,2-Trichloroethane	<0.01	0.01
Trichloroethene	<0.01	0.01
Vinyl acetate	<0.05	0.05
Xylenes (total)	<0.01	0.01

Clayton Project No. 46431-17

>C9377

Sample Matrix/Media:

Screens

Date Prepared:

03/04/93

Lab Number:

Date Analyzed:

03/04/93

Sample Identification: LAB BLANK

Analytical Method:

EPA T01 (Modified)

Additional Compounds	Concentration (ug)
Chloromethane	0.1
Chloroethene	0.07
Acetaldehyde	0.03
Butenal	0.01

Clayton Project No. 46431-17

Sample Matrix/Media: Screens Date Received: 02/08/93 Lab Number: 126823 >C9230 Date Analyzed: 02/16/93

Temperature:

30

Sample Identification: PH-F-1

Analytical Method: EPA T01 (Modified)

Volatile Compounds	Concentration*	Limit of Detection (ug)
Acetone	0.3	0.1
Benzen e	0.07	0.01
Bromodichloromethane	<0.01	0.01
Bromoform	<0.01	0.01
2-Butanone	0.1	0.1
Carbon disulfide	<0.01	0.01
Carbon tetrachloride	<0.01	0.01
Chlorobenzen e	0.01	0.01
Chloroform	<0.01	0.01
Dibromochloromethane	<0.01	0.01
1,1-Dichloroethane	<0.01	0.01
1,2-Dichloroethane	<0.01	0.01
1,1-Dichloroethene	<0.01	0.01
1,2-Dichloroethene (total)	<0.01	0.01
1,2-Dichloropropane	<0.01	0.01
cis-1,3-Dichloropropene	<0.01	0.01
trans-1,3-Dichloropropene	<0.01	0.01
Ethylbenzene	<0.01	0.01
2-Hexanone	0.05	0.05
Methylene chloride	<0.05	0.05
4-Methyl-2-pentanone	<0.05	0.05
Styrene	0.01	0.01
1,1,2,2-Tetrachloroethane	<0.01	0.01
Tetrachloroethene	<0.01	0.01
Toluene	0.02	0.01
1,1,1-Trichloroethane	<0.01	0.01
1,1,2-Trichloroethane	<0.01	0.01
Trichloroethene	<0.01	0.01
Vinyl acetate	<0.05	0.05
Xylenes (total)	<0.01	0.01

^{*} Results are blank corrected.

Clayton Project No. 46431-17

Sample Matrix/Media: Date Received: Screens 02/08/93 Lab Number: 126823 >C9230 Date Analyzed: 02/16/93 Temperature: 30

Sample Identification: PH-F-1

EPA T01 (Modified) Analytical Method:

Additional Compounds	Concentration*
Phthalic anhydride	5
C11H22 hydrocarbons	2
C12H24 hydrocarbons	2
Butenone	1
Acetic acid	0.9
Unknown compounds	0.7
Ethyl hexanol	0.4
Possible Methyl sulfone	0.2
Octanal	0.2

^{*} Results are blank corrected. Approximate Limit of Detection: 0.01 ug.

Clayton Project No. 46431-17

Sample Matrix/Media: Date Received: 02/08/93 Screens Date Analyzed: >C9229 Lab Number: 126823 02/16/93 Temperature:

Sample Identification: PH-F-1

Analytical Method: EPA T01 (Modified)

Volatile Compounds	Concentration*	Limit of Detection (ug)
Acetone	0.3	0.1
Benzene	0.16	0.01
Bromodichloromethane	<0.01	0.01
Bromoform	<0.01	0.01
2-Butanone	0.1	0.1
Carbon disulfide	<0.01	0.01
Carbon tetrachloride	<0.01	0.01
Chlorobenzene	0.01	0.01
Chloroform	<0.01	0.01
Dibromochloromethane	<0.01	0.01
1,1-Dichloroethane	<0.01	0.01
1,2-Dichloroethane	<0.01	0.01
1,1-Dichloroethene	<0.01	0.01
1,2-Dichloroethene (total)	<0.01	0.01
1,2-Dichloropropane	<0.01	0.01
cis-1,3-Dichloropropene	<0.01	0.01
trans-1,3-Dichloropropene	<0.01	0.01
Ethylbenzene	<0.01	0.01
2-Hexanone	0.11	0.05
Methylene chloride	<0.05	0.05
4-Methyl-2-pentanone	<0.05	0.05
Styrene	0.04	0.01
1,1,2,2-Tetrachloroethane	<0.01	0.01
Tetrachloroethene	<0.01	0.01
Toluene	<0.01	0.01
1,1,1-Trichloroethane	<0.01	0.01
1,1,2-Trichloroethane	<0.01	0.01
Trichloroethene	<0.01	0.01
Vinyl acetate	<0.05	0.05
Xylenes (total)	<0.01	0.01

^{*} Results are blank corrected.

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Clayton Project No. 46431-17

Sample Matrix/Media: Screens Date Received:

Lab Number: 126823 >C9229 Date Analyzed: Sample Identification: PH-F-1 Temperature:

Analytical Method: EPA T01 (Modified)

Additional Compounds	Concentration*
Nonanal	1
C11H22 hydrocarbons	1
Acetic acid	1
C12H24 hydrocarbons	0.8
C9H14O dienals	0.5
Butenone	0.5
Ethyl hexanal and Unknown hydrocarbon	0.4
Hexanal	0.4
Heptanal	0.3
Ethyl hexanol	0.3
Unknown compound	0.2
Possible Methyl sulfone	0.2

02/08/93

02/16/93

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^{*} Results are blank corrected.
Approximate Limit of Detection: 0.01 ug.